# NATIONAL EXAMS DECEMBER 2018 

## 16-Chem-B2, Environmental Engineering

## 3 hours duration <br> NOTES

1. If doubt exists as to the interpretation of any question, the candidate is urged to submit with the answer paper, a clear statement of any assumptions made.
2. This is a Closed Book Exam with a candidate prepared $8 \frac{1^{\prime \prime}}{} \times 11^{\prime \prime}$ double sided Aid-Sheet allowed.
3. Candidates may use one of two calculators, the Casio or Sharp approved models. Write the name and model designation of the calculator on the first inside left hand sheet of the exam work book.
4. Any five (5) questions constitute a complete paper. Only the first five (5) answers as they appear in your work book(s), will be marked.
5. Each question is worth a total of 20 marks with the section marks indicated in brackets () at the left margin of the question. The complete Marking Scheme is also provided on the final page. A completed exam consists of five (5) answered questions with a possible maximum score of 100 marks.

## Problem 1

Provide answers to the following questions related to engineering aspects of air pollution abatement, effluent treatment and water pollution abatement.


## Problem 2

Provide answers to the following questions related to control methods for particulates, gases and vapours. Identify three (3) differnt contaminant removal technologies [one (1) for each type of contaminant given below] and for each technology provide: (a) one (1) key design parameter that affects the performance; (b) one (1) important operational issue to ensure a high performance efficiency and (c) one (1) maintenance issue.
(i) Particulates (e.g., $\mathrm{PM}_{10}$ from quarry operation)
(ii) Gases (e.g., VOCs from spray paint booth); and
(iii) Vapours (e.g., odorous emissions from rendering plant)

## Problem 3

Provide answers to the following questions related to contaminant soil remediation and measurement techniques as applied to environmental engineering.
(iii) Identify and briefly describe the key engineering principle associated with a measurement technique used to monitor: (1) gaseous pollutants in the ambient air and (2) pathogens in treated water. As part of your description, briefly explain how calibration is performed to improve the precision of the instrument used in each case. State clearly any assumptions made in your answer.

## Problem 4

Provide answers to the following questions related to characterization of water contaminants and their measurement, biochemical oxygen demand and aeration and activated sludge process.
(i) Water quality can be measured by various characteristics needed for proper design of treatment systems including (a) turbidity, (b) dissolved oxygen and (c) nutrients. Briefly explain how each contaminant may be characterized and how it can be measured or indirectly quantified using reliable techniques.
(ii) $\mathrm{A}_{\mathrm{BOD}}^{5}$ test is conducted at standard temperature conditions using 200 mL of secondary effluent mixed with 100 mL of water. The initial DO in the mix is $6 \mathrm{mg} / \mathrm{L}$. After 5 days, the DO is $1.0 \mathrm{mg} / L$ and after 20 days the DO has stabilized at 0.03 $\mathrm{mg} / \mathrm{L}$. Assume that nitrification has been inhibited so that only $\mathrm{cBOD}_{5}$ (5-day carbonaceous biochemical oxygen demand) is being measured.
(a) Calculate the 5-day cBOD of the secondary effluent in $m g / L$; and
(b) Estimate the ultimate cBOD in $\mathrm{mg} / \mathrm{L}$.
(iii) Briefly explain three (3) main design principles to ensure the efficient performance of an aeration system used in the activated sludge wastewater treatment process.

## Problem 5

Provide answers to the following questions related to floatation, pH control, ion exchange and activated sludge process.
(i) Briefly explain the main function and an operational issue associated with each technology given below in the production of drinking water from surface water that has significant colloidal and suspended solids, a low pH and high conductivity:
(a) floatation;
(b) pH control; and
(c) ion exchange.
(ii) A conventional activated sludge plant is used to treat $200,000 \mathrm{~m}^{3} / \mathrm{d}$ of municipal wastewater. You have been asked to prepare the preliminary process design by calculating the following:
(a) The required aeration tank volume $V$ in $\mathrm{m}^{3}$ and the aeration tank hydraulic retention time $(\phi)$ in hours;
(b) the quantity of sludge to be wasted daily $\left(Q_{w}\right)$ in $\mathrm{kg} / \mathrm{d}$; and
(c) the sludge recycle ratio $\left(Q_{r} / Q_{o}\right)$.

## Use the following process information:

- Influent $B O D_{5}$ and $T S S=240 \mathrm{mg} / \mathrm{L}$;
- effluent $B O D_{5}$ and $T S S=25 \mathrm{mg} / \mathrm{L}$;
- yield coefficient, $Y=0.4$;
- decay rate, $k_{d}=0.05 \mathrm{~d}^{-1}$;
- average MLSS in the aeration tank, $X=6,000 \mathrm{mg} / \mathrm{L}$;
- waste MLSS from the clarifier, $X_{w}=9,000 \mathrm{mg} / \mathrm{L}$; and
- mean cell residence time, $\phi_{c}=12$ days;



## Problem 6

Provide answers to the following questions related to sources and dispersion of atmospheric pollutants.
A large natural gas fired power plant producing 3000 GW of power releases nitrogen dioxide $\left(\mathrm{NO}_{2}\right)$ during its operation. The $\mathrm{NO}_{2}$ is released from a 40 m stack at a rate of 20 $\mathrm{g} / \mathrm{min}$. The average wind speed is $11-15 \mathrm{~m} / \mathrm{s}$ with clear sky conditions.
(i) Provide three (3) possible engineering approaches or technologies that may be used to reduce the ground level $\mathrm{NO}_{2}$ concentration and briefly provide one (1) benefit and one (1) disadvantage associated with each engineering approach or technology.
(ii) What is the distance downwind of the plume centerline emission point at which the predicted $\mathrm{NO}_{2}$ ground-level concentration falls to less than $4 \mu \mathrm{~g} / \mathrm{m}^{3}$;
Assume an estimate of the dispersion parameters is provided by the following equations:

$$
\begin{aligned}
& \sigma_{y}=a \cdot x^{b-c \cdot \ln (x)} \\
& \sigma_{z}=d \cdot x^{e-f \cdot \ln (x)}
\end{aligned}
$$

The variables to calculate the moderated unstable dispersion parameters are to be taken from the appropriate stability class given in the table below:

| Stability Class | a | b | c | d | e | f |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A | 80 | 1.0 | -0.004 | 150 | 2.0 | 0.5 |
| B | 70 | 0.9 | -0.005 | 130 | 1.4 | 0.04 |
| C | 90 | 1.1 | -0.004 | 100 | 1.1 | 0.04 |
| D | 30 | 0.8 | -0.005 | 80 | 1.3 | -0.06 |
| E | 40 | 1.2 | -0.006 | 50 | 0.6 | -0.07 |



## Problem 7

Provide answers to the following questions related to photochemical reactions, noxious pollutants and odour control.

Photochemical smog has been identified as one of the primary causes of urban air pollution resulting in respiratory problems among the general population and other health effects among the more susceptible in our cities.
(i) Briefly explain three (3) main causes of smog and three (3) possible hard or soft engineering control methods to reduce smog production or its environmental adverse effects. For each proposed control method, provide one (1) advantage and one (1) disadvantage or challenge associated with implementing the control method.
(ii) Identify three (3) key causes of odorous emissions from the pretreatment area of a sewage treatment plant (e.g., screening room) or a furniture paint operation (e.g., paint and lamination furniture system). For the facility selected (only one), provide two (2) different control technologies that may reduce odorous emissions effectively at over $99 \%$. Briefly explain what operation and maintenance issues need to be addressed to ensure consistent operation of the emission controls.


## Marking Scheme

1. (i) 7 (ii) 6 (iii) 7 marks, 20 marks total
2. (i) 6 (ii) 7 (iii) 7 marks, 20 marks total
3. (i) 6 (ii) 6 (iii) 8 marks, 20 marks total
4. (i) 8 (ii) (a) 3, (b) 3 (iii) 6 marks, 20 marks total
5. (i) (a) 3 , (b) 4 , (c) 3 (ii) (a) 3 , (b) 4 , (c) 3 marks, 20 marks total
6. (i) 10 (ii) 10 marks, 20 marks total
7. (i) 10 (ii) 10 marks, 20 marks total
