National Exams December 2016

10-Met-B7: Physical Metallurgy of Non-Ferrous Metals and Alloys

3 hours duration

Notes:

- 1. If doubt exists as to the interpretation of any question, the candidate is urged to submit with the answer paper, a clear statement of any assumptions made.
- 2. Candidates may use one of two calculators, the Casio <u>or</u> Sharp approved models. This is a closed book exam.
- 3. Any five (5) questions constitute a complete paper (100 marks total). Only the first five questions as they appear in your answer book will be marked.
- 4. Each question is of equal value.
- 5. Most questions require an answer in essay format. Clarity and organization of the answer are important.

Question 1:

- (a) Describe in general how wrought Al-alloys are classified. How are Al casting alloys classified? (10 marks)
- (b) List the four (4) basic temper designations for Al-alloys and describe the alloy condition for each case? (10 marks)

Question 2:

- (a) The heat treatment of hardenable aluminum alloys normally involves a threestage procedure. Briefly describe the nature of the three stages and the microstructures that result following each stage. (10 marks)
- (b) The production of sheet material for structural applications requires careful control of annealing-induced transformations, either during rolling (i.e. hot working) or after cold rolling. Discuss <u>two</u> factors that control grain growth (i.e. ultimate grain size) during the annealing of aluminum alloy deformed to a specific strain. (10 marks)

Question 3:

The microstructure of as-cast copper-based alloys can be modified using one of several heat treatments. Briefly describe each of the following treatments. (Note: Your answer should consider the heat treatment procedure and the resulting microstructural changes that develop).

- (a) precipitation hardening (5 marks)
- (b) spinodal decompositon (5 marks)
- (c) homogenizing (5 marks)
- (d) stress-relieving (5 marks)

Question 4:

- (a) What are the major alloying elements added to magnesium to make magnesium alloys for engineering applications? (4 marks)
- (b) Why is grain refining not necessary for die-cast magnesium alloys? (4 marks)
- (c) What is the precipitation sequence in Mg-Al alloys, which are solution-heattreated, quenched and aged? Is the precipitation hardening effect large or small? Explain. (8 marks)
- (d) Why can't magnesium alloys be directly connected to metals like steel and copper in most engineering designs? (4 marks)

Question 5:

- (a) What alloying elements are used to make the alloy brasses? (4 marks)
- (b) What is the dezincification of brasses? What is the mechanism for this type of corrosion? (6 marks)
- (c) Why is phosphorus added to the tin bronzes? What is the chief disadvantage of these alloys when compared to the brasses? (6 marks)
- (d) What special property does aluminum provide to the alumimum bronzes to make them useful engineering alloys? (4 marks)

Question 6:

- (a) What are the principal solid-solution strengthening elements added to nickelbased superalloys? (4 marks)
- (b) How do carbides strengthen cobalt-base superalloys? (5 marks)
- (c) How does the precipitation strenghtening of the cobalt-base superalloys differ from the nickel-base superalloys? (6 marks)
- (d) What is hot corrosion of superalloys? What is one way in which hot corrosion resistance can be increased? (5 marks)

Question 7:

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- (a) What are the three types of alloy phase stabilizing systems formed in binary titanium alloys? (6 marks)
- (b) Why are beta titanium alloys more cold-formable than the alpha-titanium alloys? (5 marks)
- (c) Why must the amount of aluminum alloyed with titanium be limited to about 8%? (4 marks)
- (d) Why do transformed or partly transformed microstructures in titanium alloys have, in general, higher fracture toughness than equiaxed microstructures? (5 *marks*)

Question 8:

- (a) Give the distinctive features, limitations, and applications of: (i) refractory metals and (ii) noble metals. (10 marks)
- (b) Using a schematic phase diagram describe the difference between stoichiometric and non-stoichiometric intermetallic compounds. Choose one example of a structural intermetallic compound and explain the primary strengthening mechanism responsible for good high temperature creep resistance. (10 marks)