

NATIONAL EXAMINATIONS DECEMBER 2003

98-Nov-B1 / 98-Mec-A1 Applied Thermodynamics and Heat Transfer

3 Hours Duration

Notes :

1. If doubt exists concerning the interpretation of any question, the candidate is urged to make assumptions and clearly explain what has been assumed along with the answer to the question.
2. The examination is open book. As a consequence, candidates are permitted to make use of any textbooks, references or notes.
3. Any non-communicating calculator is permitted. However, candidates must indicate the type of calculator(s) that they have used by writing the name and model designation of the calculator(s) on the inside of the cover of the first examination book.
4. It is expected that each candidate will have copies of both a thermodynamics text and a heat transfer text in order to make use of the information presented in the tables and graphs contained.
5. The answers to five questions, either three questions from Part A and two questions from Part B or two questions from Part A and three questions from Part B, comprise a complete examination.
6. Candidates must indicate the answers that they wish to have graded on the cover of the first examination book. Otherwise the answers will be graded in the order in which they appear in the examination book(s) up to a maximum of three answers per section.
7. The answer to any question carries the same value in the grading .

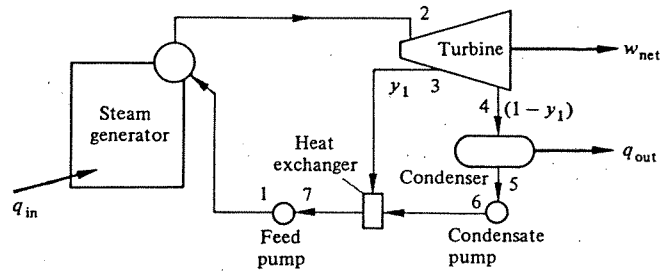
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**PART A - THERMODYNAMICS**

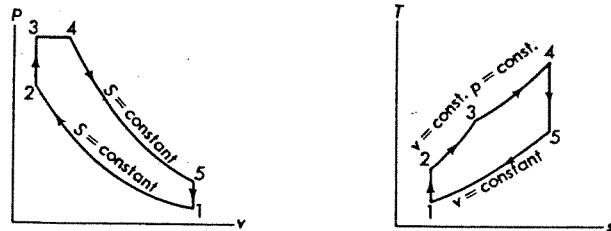
1. (a) A small capsule having a volume of  $5 \text{ cm}^3$  contains liquid water at  $7.5 \text{ MPa}$  and  $60^\circ\text{C}$ . The capsule is located inside an evacuated container having a volume of  $0.025 \text{ m}^3$ . Subsequently, the capsule is broken by some appropriate mechanism so that the water is able to evaporate and fill the large container. Calculate the final density of the water/vapour mixture, assuming that water/vapour mixture can exchange heat with the surroundings through the walls of the container. Determine the heat exchange with the surroundings.

(b) An air compressor takes air in at 1 atmosphere and  $20^\circ\text{C}$  and discharges it into a line having inside diameter of 1 cm. The air velocity in the line close to the discharge is  $7 \text{ m/s}$  and the discharge pressure is  $3.5$  atmospheres. Assuming that the compression occurs adiabatically and that the inlet air velocity is small, calculate the power required to drive the compressor.

2. Determine the efficiency and power output of a steam power plant that has turbine inlet conditions of  $6.0 \text{ MPa}$  and  $500^\circ\text{C}$ , bleeds steam to a feedwater heater at  $700 \text{ kPa}$  and exhausts to a condenser at  $25 \text{ kPa}$ . The efficiencies of the turbine and pumps are  $90\%$  and the water mass flowrate is  $63.0 \text{ kg/s}$ . Represent the processes on a  $Ts$  diagram.

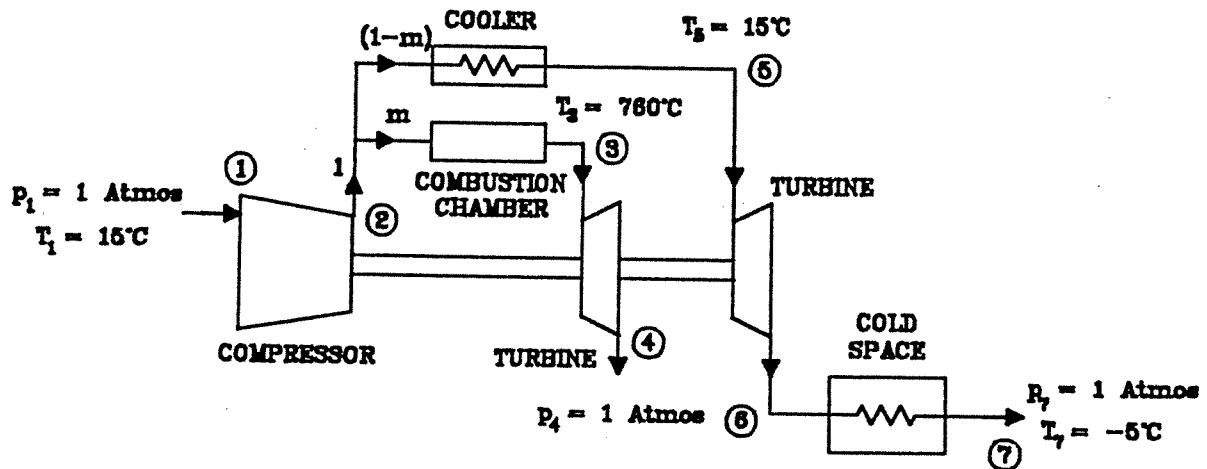


3. The schematic diagrams at the right describe the air standard dual cycle, a combination of the Otto cycle and the Diesel cycle which is a better representation of an internal combustion engine.



Heat is transferred to the working fluid at constant volume and constant pressure. Fuel injection is started before the end of the compression stroke and is completed during the early part of the return stroke. The ratios  $r_v = v_1/v_2$ ,  $r_c = v_4/v_3$  and  $r_p = p_3/p_2$  are the compression ratio, cut-off ratio and constant-volume pressure ratio respectively. In a particular air-standard dual cycle, the compression ratio is  $16.5$  and the pressure and temperature at the beginning of compression are  $101.325 \text{ kPa}$  and  $25^\circ\text{C}$ . The temperature after constant-volume combustion is  $1170^\circ\text{C}$  and the temperature after constant pressure combustion is  $1595^\circ\text{C}$ . Using the ideal gas property tables appended to the examination paper, determine the cut-off ratio, the constant volume pressure ratio, the heat added and rejected per unit mass of air and the thermal efficiency.

4.



The schematic diagram above depicts the arrangement of the components of a system proposed as a refrigeration machine. A fraction of the air mass flowrate is bled off the compressor of a simple gas turbine just before combustion, cooled to  $15^\circ\text{C}$  and then expanded through another turbine back to atmospheric pressure. The exhaust from the turbine, which is used to maintain the temperature of a cold space, finally escapes to the atmosphere at  $-5^\circ\text{C}$ . The compressor and both turbines are connected to a single shaft. The temperature of the air entering the compressor is  $15^\circ\text{C}$  and the temperature of the air entering the turbine is  $760^\circ\text{C}$ . The pressure ratio of the compressor is 5:1, the compressor efficiency is 80% and the turbine efficiency is 90%. Sketch the processes on a  $Ts$  diagram. Determine the ratio of the heat extracted from the cold space to the heat added in the combustion chamber.

#### PART B - HEAT TRANSFER

5. Heat is generated uniformly at the rate of  $q''' = 50 \text{ W/cm}^3$  in a long wire  $r_{\text{wire}} = 0.2 \text{ cm}$  having thermal conductivity  $k_{\text{wire}} = 15 \text{ W/m}^\circ\text{C}$ . The wire is sheathed in a  $\delta_{\text{ceramic}} = 0.5 \text{ cm}$  thick layer of ceramic material having thermal conductivity  $k_{\text{ceramic}} = 1.2 \text{ W/m}^\circ\text{C}$ . The ceramic insulated wire is located in quiescent air at  $T_\infty = 20^\circ\text{C}$  and the heat transfer coefficient at the ceramic surface  $h = 50 \text{ W/m}^2^\circ\text{C}$ . Determine the temperature  $T_{\text{interface}}$  at the interface between the wire and the ceramic and at the centerline of the wire  $T_{\text{centerline}}$ .

Continued on Page 4

6. Water flowing at 0.01 kg/s passes through a 1 cm diameter thin walled tube immersed in a bath of crushed ice and water such that the wall temperature may be assumed to be  $T_w = 0^\circ\text{C}$ . The inlet water temperature  $T_i = 40^\circ\text{C}$  and the outlet water temperature  $T_o = 6^\circ\text{C}$ . The pipe is long enough that the flow may be assumed to be fully developed both hydrodynamically and thermally over the length of the tube. Calculate the total rate of heat transfer between the stream of water and the cooling bath and determine the length of the tube required to effect the rate of heat transfer.
7. An oil filled heating panel has the form of a thin vertical oil filled rectangle 0.75 m high by 1.5 m in length. The heating panel convects freely by natural convection from both surfaces. The heating element inserted in the oil dissipates 690 W under steady operating conditions when located in quiescent air at  $20^\circ\text{C}$ . Determine the temperature of the surface of the heating panel.
8. Water flowing at 10 kg/s through 50 brass tubes 2.3 cm ID by 2.6 cm OD heats air flowing at 1.6 kg/s through a shell within which the tubes are located in parallel. Each tube is 6.7 m long. The heat transfer coefficients inside and outside the tubes are  $470\text{ W/m}^2\text{ }^\circ\text{C}$  and  $210\text{ W/m}^2\text{ }^\circ\text{C}$  respectively. The water enters the tubes at  $75^\circ\text{C}$  and the air enters the shell at  $15^\circ\text{C}$ . Determine the heat exchanger effectiveness, the rate of heat transfer between the water and the air and the outlet temperatures of the water and the air.

The End

## Properties of water—saturation-temperature table (SI units)\*

$v$  in  $\text{cm}^3/\text{g}$ ,  $1 \text{ cm}^3/\text{g} = 10^{-3} \text{ m}^3/\text{kg}$ ;  $h$  and  $u$  in  $\text{kJ}/\text{kg}$ ;  $s$  in  $\text{kJ}/\text{kg}\cdot\text{K}$ ;  $p$  in bars,  $1 \text{ bar} = 10^5 \text{ Pa}$

Temp. $^{\circ}\text{C}$ $T$	Press. bars $P$	Specific volume		Internal energy		Enthalpy			Entropy	
		Sat. liquid $v_f$	Sat. vapor $v_g$	Sat. liquid $u_f$	Sat. vapor $u_g$	Sat. liquid $h_f$	Evap. $h_{fg}$	Sat. vapor $h_g$	Sat. liquid $s_f$	Sat vapor $s_g$
0	0.00611	1.0002	206278	-0.03	2375.4	-0.02	2501.4	2501.3	-0.0001	9.1565
5	0.00872	1.0001	147120	20.97	2382.3	20.98	2489.6	2510.6	0.0761	9.0257
10	0.01228	1.0004	106379	42.00	2389.2	42.01	2477.7	2519.8	0.1510	8.9008
15	0.01705	1.0009	77926	62.99	2396.1	62.99	2465.9	2528.9	0.2245	8.7814
20	0.02339	1.0018	57791	83.95	2402.9	83.96	2454.1	2538.1	0.2966	8.6672
25	0.03169	1.0029	43360	104.88	2409.8	104.89	2442.3	2547.2	0.3674	8.5580
30	0.04246	1.0043	32894	125.78	2416.6	125.79	2430.5	2556.3	0.4369	8.4533
35	0.05628	1.0060	25216	146.67	2423.4	146.68	2418.6	2565.3	0.5053	8.3531
40	0.07384	1.0078	19523	167.56	2430.1	167.57	2406.7	2574.3	0.5725	8.2570
45	0.09593	1.0099	15258	188.44	2436.8	188.45	2394.8	2583.2	0.6387	8.1648
50	0.1235	1.0121	12032	209.32	2443.5	209.33	2382.7	2592.1	0.7038	8.0763
55	0.1576	1.0146	9568	230.21	2450.1	230.23	2370.7	2600.9	0.7679	7.9913
60	0.1994	1.0172	7671	251.11	2456.6	251.13	2358.5	2609.6	0.8312	7.9096
65	0.2503	1.0199	6197	272.02	2463.1	272.06	2346.2	2618.3	0.8935	7.8310
70	0.3119	1.0228	5042	292.95	2469.6	292.98	2333.8	2626.8	0.9549	7.7553
75	0.3858	1.0259	4131	313.90	2475.9	313.93	2321.4	2635.3	1.0155	7.6824
80	0.4739	1.0291	3407	334.86	2482.2	334.91	2308.8	2643.7	1.0753	7.6122
85	0.5783	1.0325	2828	355.84	2488.4	355.90	2296.0	2651.9	1.1343	7.5445
90	0.7014	1.0360	2361	376.85	2494.5	376.92	2283.2	2660.1	1.1925	7.4791
95	0.8455	1.0397	1982	397.88	2500.6	397.96	2270.2	2668.1	1.2500	7.4159
100	1.014	1.0435	1673.	418.94	2506.5	419.04	2257.0	2676.1	1.3069	7.3549
110	1.433	1.0516	1210.	461.14	2518.1	461.30	2230.2	2691.5	1.4185	7.2387
120	1.985	1.0603	891.9	503.50	2529.3	503.71	2202.6	2706.3	1.5276	7.1296
130	2.701	1.0697	668.5	546.02	2539.9	546.31	2174.2	2720.5	1.6344	7.0269
140	3.613	1.0797	508.9	588.74	2550.0	589.13	2144.7	2733.9	1.7391	6.9299
150	4.758	1.0905	392.8	631.68	2559.5	632.20	2114.3	2746.5	1.8418	6.8379
160	6.178	1.1020	307.1	674.86	2568.4	675.55	2082.6	2758.1	1.9427	6.7502
170	7.917	1.1143	242.8	718.33	2576.5	719.21	2049.5	2768.7	2.0419	6.6663
180	10.02	1.1274	194.1	762.09	2583.7	763.22	2015.0	2778.2	2.1396	6.5857
190	12.54	1.1414	156.5	806.19	2590.0	807.62	1978.8	2786.4	2.2359	6.5079
200	15.54	1.1565	127.4	850.65	2595.3	852.45	1940.7	2793.2	2.3309	6.4323
210	19.06	1.1726	104.4	895.53	2599.5	897.76	1900.7	2798.5	2.4248	6.3585
220	23.18	1.1900	86.19	940.87	2602.4	943.62	1858.5	2802.1	2.5178	6.2861
230	27.95	1.2088	71.58	986.74	2603.9	990.12	1813.8	2804.0	2.6099	6.2146
240	33.44	1.2291	59.76	1033.2	2604.0	1037.3	1766.5	2803.8	2.7015	6.1437
250	39.73	1.2512	50.13	1080.4	2602.4	1085.4	1716.2	2801.5	2.7927	6.0730
260	46.88	1.2755	42.21	1128.4	2599.0	1134.4	1662.5	2796.9	2.8838	6.0019
270	54.99	1.3023	35.64	1177.4	2593.7	1184.5	1605.2	2789.7	2.9751	5.9301
280	64.12	1.3321	30.17	1227.5	2586.1	1236.0	1543.6	2779.6	3.0668	5.8571
290	74.36	1.3656	25.57	1278.9	2576.0	1289.1	1477.1	2766.2	3.1594	5.7821
300	85.81	1.4036	21.67	1332.0	2563.0	1344.0	1404.9	2749.0	3.2534	5.7045
320	112.7	1.4988	15.49	1444.6	2525.5	1461.5	1238.6	2700.1	3.4480	5.5362
340	145.9	1.6379	10.80	1570.3	2464.6	1594.2	1027.9	2622.0	3.6594	5.3357
360	186.5	1.8925	6.945	1725.2	2351.5	1760.5	720.5	2481.0	3.9147	5.0526
374.14	220.9	3.155	3.155	2029.6	2029.6	2099.3	0	2099.3	4.4298	4.4298

\* Abridged from Keenan, J. H., F. G. Keyes, P. G. Hill, and J. G. Moore, "Steam Tables," Wiley, New York, 1969.

## Properties of water—compressed-liquid table (SI units)\*

 $v$  in  $\text{cm}^3/\text{g}$ ,  $1 \text{ cm}^3/\text{g} = 10^{-3} \text{ m}^3/\text{kg}$ ;  $h$  and  $u$  in  $\text{kJ}/\text{kg}$ ;  $s$  in  $\text{kJ}/\text{kg}\cdot\text{K}$ 

Temp. C	$v$	$u$	$h$	$s$	$v$	$u$	$h$	$s$
	2.5 MPa (223.99°C)				5.0 MPa (263.99°C)			
20	1.0006	83.80	86.30	0.2961	0.9995	83.65	88.65	0.2956
40	1.0067	167.25	169.77	0.5715	1.0056	166.95	171.97	0.5705
80	1.0280	334.29	336.86	1.0737	1.0268	333.72	338.85	1.0720
120	1.0590	502.68	505.33	1.5255	1.0576	501.80	507.09	1.5233
160	1.1006	673.90	676.65	1.9404	1.0988	672.62	678.12	1.9375
200	1.1555	849.9	852.8	2.3294	1.1530	848.1	848.1	2.3255
220	1.1898	940.7	943.7	2.5174	1.1866	938.4	944.4	2.5128
Sat.	1.1973	959.1	962.1	2.5546	1.2859	1147.8	1154.2	2.9202
7.5 MPa (290.59°C)				10.0 MPa (311.06°C)				
20	0.9984	83.50	90.99	0.2950	0.9972	83.36	93.33	0.2945
40	1.0045	166.64	174.18	0.5696	1.0034	166.35	176.38	0.5686
80	1.0256	333.15	340.84	1.0704	1.0245	332.59	342.83	1.0688
100	1.0397	416.81	424.62	1.3011	1.0385	416.12	426.50	1.2992
140	1.0752	585.72	593.78	1.7317	1.0737	584.68	595.42	1.7292
180	1.1219	758.13	766.55	2.1308	1.1199	756.65	767.84	2.1275
220	1.1835	936.2	945.1	2.5083	1.1805	934.1	945.9	2.5039
260	1.2696	1124.4	1134.0	2.8763	1.2645	1121.1	1133.7	2.8699
Sat.	1.3677	1282.0	1292.2	3.1649	1.4524	1393.0	1407.6	3.3596
15.0 MPa (342.24°C)				20.0 MPa (365.81°C)				
20	0.9950	83.06	97.99	0.2934	0.9928	82.77	102.62	0.2923
40	1.0013	165.76	180.78	0.5666	0.9992	165.17	185.16	0.5646
100	1.0361	414.75	430.28	1.2955	1.0337	413.39	434.06	1.2917
180	1.1159	753.76	770.50	2.1210	1.1120	750.95	773.20	2.1147
220	1.1748	929.9	947.5	2.4953	1.1693	925.9	949.3	2.4870
260	1.2550	1114.6	1133.4	2.8576	1.2462	1108.6	1133.5	2.8459
300	1.3770	1316.6	1337.3	3.2260	1.3596	1306.1	1333.3	3.2071
Sat.	1.6581	1585.6	1610.5	3.6848	2.036	1785.6	1826.3	4.0139
25.0 MPa				30.0 MPa				
20	0.9907	82.47	107.24	0.2911	0.9886	82.17	111.84	0.2899
40	0.9971	164.60	189.52	0.5626	0.9951	164.04	193.89	0.5607
100	1.0313	412.08	437.85	1.2881	1.0290	410.78	441.66	1.2844
200	1.1344	834.5	862.8	2.2961	1.1302	831.4	865.3	2.2893
300	1.3442	1296.6	1330.2	3.1900	1.3304	1287.9	1327.8	3.1741

\* Abridged from Keenan, J. H., F. G. Keyes, P. G. Hill, and J. G. Moore, "Steam Tables," Wiley, New York, 1969.

# Ideal Gas Properties of Air

Dec. 2003

$T(K), h$ and $u(kJ/kg), s^{\circ}(kJ/kg \cdot K)$											
$T$	$h$	$P_r$	$u$	$u_r$	$s^{\circ}$	$T$	$h$	$P_r$	$u$	$u_r$	$s^{\circ}$
200	199.97	0.3363	142.56	1707.	1.29559	750	767.29	37.35	551.99	57.63	2.64737
210	209.97	0.3987	149.69	1512.	1.34444	760	778.18	39.27	560.01	55.54	2.66176
220	219.97	0.4690	156.82	1346.	1.39105	770	789.11	41.31	568.07	53.39	2.67595
230	230.02	0.5477	164.00	1205.	1.43557	780	800.03	43.35	576.12	51.64	2.69013
240	240.02	0.6355	171.13	1084.	1.47824	790	810.99	45.55	584.21	49.86	2.70400
250	250.05	0.7329	178.28	979.	1.51917	800	821.95	47.75	592.30	48.08	2.71787
260	260.09	0.8405	185.45	887.8	1.55848	820	843.98	52.59	608.59	44.84	2.74504
270	270.11	0.9590	192.60	808.0	1.59634	840	866.08	57.60	624.95	41.85	2.77170
280	280.13	1.0889	199.75	738.0	1.63279	860	888.27	63.09	641.40	39.12	2.79783
285	285.14	1.1584	203.33	706.1	1.65055	880	910.56	68.98	657.95	36.61	2.82344
290	290.16	1.2311	206.91	676.1	1.66802	900	932.93	75.29	674.58	34.31	2.84856
295	295.17	1.3068	210.49	647.9	1.68515	920	955.38	82.05	691.28	32.18	2.87324
300	300.19	1.3860	214.07	621.2	1.70203	940	977.92	89.28	708.08	30.22	2.89748
305	305.22	1.4686	217.67	596.0	1.71865	960	1000.55	97.00	725.02	28.40	2.92128
310	310.24	1.5546	221.25	572.3	1.73498	980	1023.25	105.2	741.98	26.73	2.94468
315	315.27	1.6442	224.85	549.8	1.75106	1000	1046.04	114.0	758.94	25.17	2.96770
320	320.29	1.7375	228.42	528.6	1.76690	1020	1068.89	123.4	776.10	23.72	2.99034
325	325.31	1.8345	232.02	508.4	1.78249	1040	1091.85	133.3	793.36	22.39	3.01260
330	330.34	1.9352	235.61	489.4	1.79783	1060	1114.86	143.9	810.62	21.14	3.03449
340	340.42	2.149	242.82	454.1	1.82790	1080	1137.89	155.2	827.88	19.98	3.05608
350	350.49	2.379	250.02	422.2	1.85708	1100	1161.07	167.1	845.33	18.896	3.07732
360	360.58	2.626	257.24	393.4	1.88543	1120	1184.28	179.7	862.79	17.886	3.09825
370	370.67	2.892	264.46	367.2	1.91313	1140	1207.57	193.1	880.35	16.946	3.11883
380	380.77	3.176	271.69	343.4	1.94001	1160	1230.92	207.2	897.91	16.064	3.13916
390	390.88	3.481	278.93	321.5	1.96633	1180	1254.34	222.2	915.57	15.241	3.15916
400	400.98	3.806	286.16	301.6	1.99194	1200	1277.79	238.0	933.33	14.470	3.17888
410	411.12	4.153	293.43	283.3	2.01699	1220	1301.31	254.7	951.09	13.747	3.19834
420	421.26	4.522	300.69	266.6	2.04142	1240	1324.93	272.3	968.95	13.069	3.21751
430	431.43	4.915	307.99	251.1	2.06533	1260	1348.55	290.8	986.90	12.435	3.23638
440	441.61	5.332	315.30	236.8	2.08870	1280	1372.24	310.4	1004.76	11.835	3.25510
450	451.80	5.775	322.62	223.6	2.11161	1300	1395.97	330.9	1022.82	11.275	3.27345
460	462.02	6.245	329.97	211.4	2.13407	1320	1419.76	352.5	1040.88	10.747	3.29160
470	472.24	6.742	337.32	200.1	2.15604	1340	1443.60	375.3	1058.94	10.247	3.30959
480	482.49	7.268	344.70	189.5	2.17760	1360	1467.49	399.1	1077.10	9.780	3.32724
490	492.74	7.824	352.08	179.7	2.19876	1380	1491.44	424.2	1095.26	9.337	3.34474
500	503.02	8.411	359.49	170.6	2.21952	1400	1515.42	450.5	1113.52	8.919	3.36200
510	513.32	9.031	366.92	162.1	2.23993	1420	1539.44	478.0	1131.77	8.526	3.37901
520	523.63	9.684	374.36	154.1	2.25997	1440	1563.51	506.9	1150.13	8.153	3.39586
530	533.98	10.37	381.84	146.7	2.27967	1460	1587.63	537.1	1168.49	7.801	3.41247
540	544.35	11.10	389.34	139.7	2.29906	1480	1611.79	568.8	1186.95	7.468	3.42892
550	554.74	11.86	396.86	133.1	2.31809	1500	1635.97	601.9	1205.41	7.152	3.44516
560	565.17	12.66	404.42	127.0	2.33685	1520	1660.23	636.5	1223.87	6.854	3.46120
570	575.59	13.50	411.97	121.2	2.35531	1540	1684.51	672.8	1242.43	6.569	3.47712
580	586.04	14.38	419.55	115.7	2.37348	1560	1708.82	710.5	1260.99	6.301	3.49276
590	596.52	15.31	427.15	110.6	2.39140	1580	1733.17	750.0	1279.65	6.046	3.50829
600	607.02	16.28	434.78	105.8	2.40902	1600	1757.57	791.2	1298.30	5.804	3.52364
610	617.53	17.30	442.42	101.2	2.42644	1620	1782.00	834.1	1316.96	5.574	3.53879
620	628.07	18.36	450.09	96.92	2.44356	1640	1806.46	878.9	1335.72	5.355	3.55381
630	638.63	19.84	457.78	92.84	2.46048	1660	1830.96	925.6	1354.48	5.147	3.56867
640	649.22	20.64	465.50	88.99	2.47716	1680	1855.50	974.2	1373.24	4.949	3.58335
650	659.84	21.86	473.25	85.34	2.49364	1700	1880.1	1025	1392.7	4.761	3.5979
660	670.47	23.13	481.01	81.89	2.50985	1750	1941.6	1161	1439.8	4.328	3.6336
670	681.14	24.46	488.81	78.61	2.52589	1800	2003.3	1310	1487.2	3.944	3.6684
680	691.82	25.85	496.62	75.50	2.54175	1850	2065.3	1475	1534.9	3.601	3.7023
690	702.52	27.29	504.45	72.56	2.55731	1900	2127.4	1655	1582.6	3.295	3.7354
700	713.27	28.80	512.33	69.76	2.57277	1950	2189.7	1852	1630.6	3.022	3.7677
710	724.04	30.38	520.23	67.07	2.58810	2000	2252.1	2068	1678.7	2.776	3.7994
720	734.82	32.02	528.14	64.53	2.60319	2050	2314.6	2303	1726.8	2.555	3.8303
730	745.62	33.72	536.07	62.13	2.61803	2100	2377.4	2559	1775.3	2.356	3.8605
740	756.44	35.50	544.02	59.82	2.63280	2150	2440.3	2837	1823.8	2.175	3.8901